<u>Chemi-ionization/recombination processes</u> <u>in the broad-line region of AGNs</u>

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Work done by Team: prof. Dimitrijevic, Ignjatovic...Institute of physics Belgrade&AOB

Also, we wont to collaborate with experts in AGNs who can help us. Prof. Ilic, Popovic...

It is of interest to investigate the influence of the <u>atomic processes</u> in <u>dense</u> <u>parts of BLR clouds</u> and to provide the data useful for modelling and investigations of such layers.

The aim of this investigation is to study the <u>physics</u> of AGN, namely to investigate the atomic processes (collisional atom - Rydberg atom i.e. chemiionization/ recombination and also n-n' mixing processes) and <u>revise their</u> <u>role</u>.

In addition, this is particularly significant to better estimate the <u>hydrogen</u> <u>Balmer lines fluxes</u>, which usage for <u>effective temperature diagnostics</u> in astrophysical plasma is <u>limited</u> by errors from the line formation models and <u>uncertainties in used atomic data of hydrogen atom and inelastic collisions</u>.

We wont to <u>find</u> out <u>at what plasma conditions</u> certain <u>atomic processes</u> <u>become important</u> and could explain the existence of AGN regions with such characteristics, and could be used for future diagnostics, numerical simulations and modelling.

<u>Chemi-ionization/recombination processes in the solar</u> <u>photosphere</u>

Short Intro: Solar photosphere and M red dwarf atmosphere, before AGN BLR talk.

Few years ago we started investigations of chemi-ionization/recombination processes and their influence in the solar photosphere.

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CHEMI-IONIZATION IN SOLAR PHOTOSPHERE: INFLUENCE ON THE HYDROGEN ATOM EXCITED STATES POPULATION

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ABSTRACT

In this paper, the influence of chemi-ionization processes in $H^*(n \ge 2) + H(1s)$ collisions, as well as the influence of inverse chemi-recombination processes on hydrogen atom excited-state populations in solar photosphere, are compared with the influence of concurrent electron-atom and electron-ion ionization and recombination processes. It has been found that the considered chemi-ionization/recombination processes dominate over the relevant concurrent processes in almost the whole solar photosphere. Thus, it is shown that these processes and their importance for the non-local thermodynamic equilibrium modeling of the solar atmosphere should be investigated further. In Mihajlov et al. (2011) we investigated chemi-ionization/recombination processes in $H^*(n) + H(1s)$ collisions and their influence on the populations of hydrogen Rydberg atoms and free electrons in weakly ionized layers of the solar photosphere and the lower chromosphere.

We analyse chemi-ionization processes (collisional ionization): 2 channels

 $H^*(n) + H(1s) \Rightarrow H_2^+ + e$, Associative chemi-ionization channel i.e. creation of H_2^+ molecular ion.

 $H^*(n) + H(1s) \Rightarrow H(1s) + H^+ + e$, Non-associative ionization channel, penning ionization channel

Inverse chemi-recombination processes:

$$H_2^+ + e \Rightarrow H^*(n) + H(1s),$$

 $H(1s) + H^+ + e \Rightarrow H^*(n) + H(1s),$

We <u>compare</u> with the corresponding electron-atom and other concurrent collision processes

$$H^{*}(n) + e \Rightarrow H^{+} + 2e,$$

$$H^{+} + 2e \Rightarrow H^{*}(n) + e,$$

$$H^{+} + e \Rightarrow H^{*}(n) + \varepsilon_{\lambda},$$

Ground state

$$A^{*}(n) + A \Longrightarrow A_{2}^{+} + e \qquad (4)$$
$$A^{*}(n) + A \Longrightarrow A + A^{+} + e \qquad (5)$$

Mechanism of collisional processes (4) and (5) is described in details e.g. in review paper Mihajlov et al. 2012.

 $A^*(n) + A$ as <u>collisional complex</u> is represented as $[A^+ + A(1s)] + e$ i.e. <u>system of quasi</u> <u>molecule and electron</u>.







Using method described e.g. in Mihajlov et al. (2012), we <u>calculated cross sections</u> and <u>rate coefficients</u> for the <u>conditions</u> that exists in weakly ionized layers of the <u>solar</u> <u>photosphere</u> and the lower chromosphere.

 $H^*(n) + H(1s) \Rightarrow H_2^+ + e$, ionization channel a) $H^*(n) + H(1s) \Rightarrow H(1s) + H^+ + e$, ionization channel b)

Cross sections for channels a) and b)
$$\sigma_{ci}^{(a,b)}(n, E) = 2\pi \int_{0}^{\rho_{max}^{(a,b)}(E)} P_{ci}^{(a,b)}(n, \rho, E)\rho d\rho,$$

Rate coefficient for channels a) and b)
$$K_{ci}^{(a,b)}(n,T) = \int_{E_{min}^{(a,b)}(n)}^{E_{max}} v\sigma_{ci}^{(a,b)}(n,E)f(v;T)dv,$$

Total rate coefficient

$$K_{\rm ci}(n, T) = K_{\rm ci}^{(a)}(n, T) + K_{\rm ci}^{(b)}(n, T),$$

These data are <u>input parameters</u> needed for <u>modeling</u>.

<i>T</i> (K)	n											
	2	3	4	5	6	7	8					
4000	0.150E-11	0.619E-09	0.126E-08	0.576E-09	0.554E-09	0.463E-09	0.366E-09					
4250	0.202E-11	0.549E-09	0.106E-08	0.617E-09	0.583E-09	0.482E-09	0.378E-09					
4500	0.260E-11	0.501E-09	0.900E-09	0.656E-09	0.611E-09	0.500E-09	0.389E-09					
4750	0.324E-11	0.488E-09	0.833E-09	0.694E-09	0.637E-09	0.517E-09	0.400E-09					
5000	0.403E-11	0.495E-09	0.815E-09	0.730E-09	0.662E-09	0.533E-09	0.410E-09					
5250	0.504E-11	0.501E-09	0.800E-09	0.765E-09	0.686E-09	0.548E-09	0.420E-09					
5500	0.623E-11	0.500E-09	0.782E-09	0.799E-09	0.709E-09	0.563E-09	0.428E-09					
5750	0.756E-11	0.493E-09	0.764E-09	0.832E-09	0.731E-09	0.576E-09	0.437E-09					
6000	0.909E-11	0.490E-09	0.757E-09	0.864E-09	0.752E-09	0.589E-09	0.445E-09					
6250	0.108E-10	0.502E-09	0.766E-09	0.895E-09	0.772E-09	0.602E-09	0.453E-09					
6500	0.128E-10	0.519E-09	0.783E-09	0.924E-09	0.791E-09	0.613E-09	0.460E-09					
7000	0.175E-10	0.540E-09	0.808E-09	0.981E-09	0.827E-09	0.635E-09	0.473E-09					
7500	0.232E-10	0.574E-09	0.848E-09	0.103E-08	0.860E-09	0.655E-09	0.485E-09					
8000	0.300E-10	0.609E-09	0.891E-09	0.108E-08	0.892E-09	0.674E-09	0.497E-09					
8500	0.380E-10	0.650E-09	0.939E-09	0.113E-08	0.920E-09	0.691E-09	0.507E-09					
9000	0.470E-10	0.688E-09	0.986E-09	0.118E-08	0.948E-09	0.707E-09	0.516E-09					
9500	0.574E-10	0.733E-09	0.104E-08	0.122E-08	0.973E-09	0.722E-09	0.525E-09					
10000	0.689E-10	0.787E-09	0.109E-08	0.126E-08	0.997E-09	0.736E-09	0.533E-09					

 Table 1

 Calculated Values of Coefficient K_{ci} (cm³ s⁻¹) as a Function of n and T

<i>T</i> (K)				n			
	2	3	4	5	6	7	8
4000	0.190E-27	0.732E-27	0.390E-27	0.114E-27	0.977E-28	0.831E-28	0.709E-28
4250	0.130E-27	0.458E-27	0.257E-27	0.102E-27	0.880E-28	0.753E-28	0.645E-28
4500	0.918E-28	0.305E-27	0.177E-27	0.914E-28	0.799E-28	0.688E-28	0.591E-28
4750	0.666E-28	0.223E-27	0.135E-27	0.828E-28	0.730E-28	0.631E-28	0.544E-28
5000	0.506E-28	0.174E-27	0.110E-27	0.755E-28	0.671E-28	0.582E-28	0.503E-28
5250	0.403E-28	0.138E-27	0.912E-28	0.693E-28	0.619E-28	0.540E-28	0.467E-28
5500	0.331E-28	0.111E-27	0.763E-28	0.639E-28	0.575E-28	0.502E-28	0.436E-28
5750	0.275E-28	0.889E-28	0.645E-28	0.592E-28	0.535E-28	0.469E-28	0.407E-28
6000	0.233E-28	0.731E-28	0.558E-28	0.551E-28	0.500E-28	0.440E-28	0.382E-28
6250	0.201E-28	0.627E-28	0.498E-28	0.514E-28	0.469E-28	0.413E-28	0.360E-28
6500	0.176E-28	0.548E-28	0.451E-28	0.482E-28	0.441E-28	0.389E-28	0.339E-28
7000	0.139E-28	0.421E-28	0.374E-28	0.427E-28	0.393E-28	0.348E-28	0.304E-28
7500	0.114E-28	0.341E-28	0.322E-28	0.382E-28	0.354E-28	0.314E-28	0.275E-28
8000	0.964E-29	0.284E-28	0.283E-28	0.345E-28	0.321E-28	0.286E-28	0.250E-28
8500	0.834E-29	0.243E-28	0.253E-28	0.314E-28	0.293E-28	0.261E-28	0.229E-28
9000	0.731E-29	0.211E-28	0.229E-28	0.287E-28	0.269E-28	0.240E-28	0.211E-28
9500	0.654E-29	0.187E-28	0.209E-28	0.264E-28	0.248E-28	0.222E-28	0.195E-28
10000	0.590E-29	0.169E-28	0.194E-28	0.245E-28	0.230E-28	0.206E-28	0.181E-28

 Table 2

 Calculated Values of Recombination Coefficient K_{cr} (cm⁶ s⁻¹) as a Function of n and T

The <u>first conclusion</u> from the paper Mihajlov et al. (2011) relates to relative contribution of partial chemi-ionization/recombination processes i.e. channels a) and b) for given n and T



<u>Second conclusion</u> relates to the <u>comparison</u> with the corresponding electron-atom collision processes

F_{içea}(n)

Comparison of Fluxes of the Considered Processes

 $I_{ci}(n, T) = K_{ci}(n, T) \cdot N_n N_1,$ $I_{cr}(n, T) = K_{cr}(n, T) \cdot N_1 N_i N_e,$ $I_{i;ca}(n, T) = K_{ca}(n, T) \cdot N_n N_e,$ $I_{r;cci}(n, T) = K_{cci}(n, T) \cdot N_i N_e N_e,$ $I_{r;cb}(n, T) = K_{pb}(n, T) \cdot N_i N_e,$

<u>Quantity F(n,T) i.e.</u> ratio of fluxes. Partial, for every n

$$F_i(n,T) = \frac{I_{\rm ci}(n,T)}{I_{\rm i;ca}(n,T)} = \frac{K_{\rm ci}(n,T)}{K_{\rm ca}(n,T)} \cdot N_1 N_{\rm e},$$

Total, for the whole block of the excited hydrogen atom states with $2 \le n \le 8$

$$F_{i,\text{ea};2-8}(T) = \frac{\sum_{n=2}^{8} I_{\text{ci}}(n, T)}{\sum_{n=2}^{8} I_{i;\text{ea}}(n, T)}$$

It has been demonstrated in Mihajlov et al. (2011) that chemiionization/ recombination processes in H(n) + H(1s) collisions, for the principal quantum number n>2, must have <u>significant influence in</u> <u>comparison</u> with the corresponding electron-atom collision processes <u>on the populations</u> of <u>hydrogen Rydberg atoms</u> and <u>electrons</u> in weakly ionized layers of the solar photosphere and the lower chromosphere, and that they have to be included in modelling and investigation of solar plasma, especially in the <u>region</u> of the <u>temperature minimum</u> in the Solar photosphere. And should <u>influence</u> on the <u>atomic spectral line shapes</u>.



Figure 4. Behavior of the quantity $F_{i;ea}(2; 8)$ given by Equation (29), as a function of height *h*.

functions of height (h) in solar photosphere

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Figure 4. Behavior of the quantity $F_{i;ea}(2; 8)$ given by Equation (29), as a function of height *h*.



Figure 1. Basic plasma parameters, for the solar model of Vernazza et al. (1981), as a function of height *h*.

Next result:

<u>M red dwarf atmosphere with the</u> effective temperature T_{eff} = 3800 K



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Rydberg atoms in astrophysics

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The results suggest that the chemiionization/recombination processes due to their influence on the excited state populations and the free electron density, also should <u>influence</u> on the atomic <u>spectral line shapes</u>.

This <u>assumption</u> is <u>confirmed</u> by the Figs, which show the profiles of some of hydrogen spectral lines in the M red stars calculated with and without these processes using <u>PHOENIX</u> code.



Figure 13 (from [26]) show the line profiles of H_{α} , H_{δ} , H_{ϵ} Pa_{ϵ} with and without inclusion of processes (1). Profiles are synthesized with PHOENIX code with Stark broadening contribution calculated using tables from [40] for Stark broadening of hydrogen lines (linear Stark effect). Lineshape changes, especially in the wings, show the influence of the electron density change having a direct influence on the Stark broadening of hydrogen lines.

Let's go back to the chemi-ionization/recombination atomic processes in the <u>AGNs broad-line region</u>



Out main idea

In AGN, especially in the region of the moderately ionized layers of dense parts of the BLR clouds <u>plasma conditions are closer to stellar</u> atmospheres than to photoionized nebulae (Osterbrock 1989).

Consequently, it is of interest to investigate the influence of the mentioned processes in dense parts of BLR clouds and to provide the data on the corresponding rate coefficients useful for modelling and investigations of such layers. That is conlusion of papers Netzer 1990; Jogee 2006; Osterbrock & Ferland 2006 :

In order to <u>develop</u> and <u>improve</u> <u>diagnostic methods</u> needed for the estimation of the physical conditions in the particular parts of active galactic nuclei (AGNs), the <u>investigation</u> of the influence of various relevant <u>atomic and molecular collisional processes</u> is <u>needed</u>.

we started to check the literature. problems

Paper of Mercè Crosas and Jon C. Hydrogen molecules in Weisheit, quasar broad-line regions, MNRAS, 1993, 262, 359.

they report uncertainties of the rate coefficients due to hydrogen collisions in almost all cases.

CI: No rate coefficients for higher *n* (*n*>3) and no rate coefficients for second non-associative channel.

For the conditions 10⁴ - 10¹⁰ cm⁻³ they concluded that the influence of the associative chemi-ionization processes is negligible in BLR clouds.

However, we assume that in very dense weakly ionized regions with > 10¹⁰ cm⁻³ (if exists), these chemi ionization/ recombination processes could be important and could change the optical characteristics. 12th SCSLSA

Table 2. Collisional reactions.

Reaction	Rate Coefficient (cm ³ /s)		Ref
1. $\mathrm{H}^+ + e^- \rightarrow \mathrm{H} + h\nu$	$K_1 = 1.59 \times 10^{-13} T_4^{-0.5} e^{-\tau_{LL}} + 2.6 \times 10^{-13}$	T_4^{-0.85}	1
2. $H+e^- \rightarrow H^- + h\nu$	${\rm K_2} = 2.65 \times 10^{-15} {\rm T_4} + 1.22 \times 10^{-15}$		2
3. $H^- + H \rightarrow H_2 + e^-$	${\rm K}_{3}=2.7\times 10^{-9}$		3
4. $H^- + H^+ \rightarrow H + H$	${\rm K_4}=7.0\times 10^{-9}{\rm T_4^{-0.5}}$		4
5. $H^- + H_2^+ \rightarrow H + H + H$ $\rightarrow H_2 + H$	$K_5 = 5.0 \times 10^{-8} T_4^{-0.5}$		4
6. $\mathrm{H^+} + \mathrm{H} \rightarrow \mathrm{H_2^+} + h\nu$	${\rm K}_6 = 2.9 \times 10^{-16} {\rm T}_4^{-1.8},$	T < 6700 K	5
	$\mathrm{K_6} = 5.8 \times 10^{-16} (\mathrm{T_4/5.6})^{-0.66 \log 10 (\mathrm{T_4/5.6})},$	T > 6700K	5
7. $H_2^+ + H \rightarrow H_2 + H^+$	$K_7 = 6.4 \times 10^{-10}$		6
8. $H_2 + H^+ \rightarrow H_2^+ + H$	$K_8 = 2.4 \times 10^{-9} e^{-2.12/T_4}$		7
9. $H_2^+ + e^- \rightarrow H^+ + H + e^-$	$K_{9} = 2.0 \times 10^{-8}$		8
10. $H_2 + e^- \rightarrow 2H + e^-$	$K_{10} = 1.1 \times 10^{-8} e^{-10.2/T_4} T_4^{0.35}$		5
11. $H_2 + e^- \rightarrow H^- + H$	$K_{11} = 9.69 \times 10^{-13} e^{-11.323/(\ln 10^4 T_4 - 7.28)}$		5
12. $H+H+H \rightarrow H_2 + H$	$K_{12} = 5.5 \times 10^{-33} T_4^{-1}$		5
13. $H_2 + H \rightarrow H + H + H$	$K_{13}=6.53\times 10^{-9}e^{-5.24/T_4}$		5
14. $H_2 + H + H \rightarrow H_2 + H_2$	$K_{13} = K_{10}/8$		5
15. $H_2 + H_2 \rightarrow H_2 + H + H$	$K_{15} = 11.3 \times 10^{-9} e^{-5.33/T_4}$		5
16. $H(n=1)+H^{\bullet}(n=2) \rightarrow H_2^{\bullet} \rightarrow H_2 + h\nu$	$\rm K_{16} = 5.0 \times 10^{-14}$		9
17. $H(n=1)+H^*(n=2) \rightarrow H_2^+ + e^-$	$\mathrm{K_{17}=8.73\times10^{-12}T_4^{0.95}},$	T < 5000 K	10
	$\mathrm{K_{17}=2.9\times10^{-11}T_{4}^{2.69}},$	T > 5000 K	10
18. $H(n=1)+H^{\bullet}(n=3) \rightarrow H_2^+ + e^-$	$K_{18} = 4.42 \times 10^{-11} T_4^{0.95} e^{0.87/T_4}, \label{eq:K18}$	T < 5000 K	11
	$K_{18} = 1.47 \times 10^{-10} T_4^{2.69} e^{0.87/T_4},$	T > 5000 K	11

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For example in Marziani et al. (2011), Negrete et al. (2012) and Marziani et al. (2015), hydrogen atom density $7.00 \le \log n_{\rm H} \le 14.00$ has been used for various simulations in BLR with the code CLOUDY (Ferland et al. 2013). Higher densities exists.

On the other hand, the illuminated surface of the BLR clouds is highly ionized, but if they are sufficiently large the <u>temperature may decrease</u> to the much lower values, e.g. up to around <u>2000 K</u>, as was taken in Crosas & Weisheit (1993), where gas is weakly ionized with large amount of hydrogen molecules (Crosas & Weisheit 1993).

If very dense weakly ionized regions exist, these <u>chemi-ionization/</u> <u>recombination processes</u> could be <u>important</u> and could change the optical characteristics. there is a need for accurate data

$$\sigma_{\rm ci}^{(a,b)}(n,E) = 2\pi \int_0^{\rho_{\rm max}^{(a,b)}(E)} P_{\rm ci}^{(a,b)}(n,\rho,E) \rho d\rho,$$

$$K_{\rm ci}^{(a,b)}(n,T) = \int_{E_{\rm min}^{(a,b)}(n)}^{E_{\rm max}} v \sigma_{\rm ci}^{(a,b)}(n,E) f(v;T) dv,$$

$$K_{\rm ci}(n, T) = K_{\rm ci}^{(a)}(n, T) + K_{\rm ci}^{(b)}(n, T),$$



Figure 2. Plot of collisional ionization H(n) + H(1s) rate coefficients for selected excited states (n = 4, 7, 10). The black lines are the data analysed in Barklem (2007) for non-associative channel (4), where A = H. The data from

Literature: the <u>uncertainties</u> of the rate coefficients in hydrogen collision. Differences, almost few order of magnitudes from different sources.



Figure 1. (a) The surface plot of the partial cross-section $\sigma_{ci}^{(a)}(n, E)$ equation (10) of the chemi-ionization processes (3), i.e. associative ionization channel. (b) The surface plot of the partial cross-sections $\sigma_{ci}^{(b)}(n, E)$ equation (10) of the chemi-ionization processes (4), i.e. non-associative ionization channel.

T/n	2	3	4	5	6	7	8	9	10
3000	9.20E-13	7.32E-10	1.61E-09	3.99E-10	4.21E-10	3.73E-10	3.06E-10	2.44E-10	1.92E-10
3500	1.10E-12	6.90E-10	1.57E-09	4.90E-10	4.91E-10	4.21E-10	3.38E-10	2.65E-10	2.06E-10
4000	1.50E-12	6.19E-10	1.26E-09	5.76E-10	5.54E-10	4.63E-10	3.66E-10	2.83E-10	2.17E-10
4500	2.60E-12	5.01E-10	9.00E-10	6.56E-10	6.11E-10	5.00E-10	3.89E-10	2.98E-10	2.28E-10
5000	4.03E-12	4.95E-10	8.15E-10	7.30E-10	6.62E-10	5.33E-10	4.10E-10	3.11E-10	2.36E-10
5500	6.23E-12	5.00E-10	7.82E-10	7.99E-10	7.09E-10	5.63E-10	4.28E-10	3.23E-10	2.44E-10
6000	9.09E-12	4.90E-10	7.57E-10	8.64E-10	7.52E-10	5.89E-10	4.45E-10	3.33E-10	2.51E-10
6500	1.28E-11	5.19E-10	7.83E-10	9.24E-10	7.91E-10	6.13E-10	4.60E-10	3.42E-10	2.56E-10
7000	1.75E-11	5.40E-10	8.08E-10	9.81E-10	8.27E-10	6.35E-10	4.73E-10	3.51E-10	2.62E-10
7500	2.32E-11	5.74E-10	8.48E-10	1.03E-09	8.60E-10	6.55E-10	4.85E-10	3.58E-10	2.67E-10
8000	3.00E-11	6.09E-10	8.91E-10	1.08E-09	8.92E-10	6.74E-10	4.97E-10	3.65E-10	2.71E-10
8500	3.80E-11	6.50E-10	9.39E-10	1.13E-09	9.20E-10	6.91E-10	5.07E-10	3.72E-10	2.75E-10
9000	4.70E-11	6.88E-10	9.86E-10	1.18E-09	9.48E-10	7.07E-10	5.16E-10	3.77E-10	2.79E-10
9500	5.74E-11	7.33E-10	1.04E-09	1.22E-09	9.73E-10	7.22E-10	5.25E-10	3.83E-10	2.82E-10
10000	6 80F 11	7 87E 10	1.00₽.00	1.96F 00	0.07E 10	7 26F 10	K 22E 10	3.88E-10	2.85E-10
11000	9.47E-11	7.49E-10	1.31E-09	1.33E-09	1.04E-09	7.61E-10	5.48E-10	3.97E-10	2.91E-10
12000	1.25E-10	8.39E-10	1.41E-09	1.40E-09	1.08E-09	7.83E-10	5.61E-10	4.05E-10	2.96E-10
13000	1.59E-10	9.29E-10	1.51E-09	1.46E-09	1.11E-09	8.03E-10	5.73E-10	4.12E-10	3.00E-10
14000	1.95E-10	1.02E-09	1.60E-09	1.52E-09	1.15E-09	8.21E-10	5.83E-10	4.18E-10	3.04E-10
15000	2.35E-10	1.11E-09	1.68E-09	1.57E-09	1.18E-09	8.37E-10	5.92E-10	4.23E-10	3.08E-10
16000	2.76E-10	1.20E-09	1.77E-09	1.62E-09	1.20E-09	8.52E-10	6.01E-10	4.29E-10	3.11E-10
17000	3.19E-10	1.28E-09	1.84E-09	1.66E-09	1.23E-09	8.65E-10	6.08E-10	4.33E-10	3.14E-10
18000	3.63E-10	1.37E-09	1.92E-09	1.71E-09	1.25E-09	8.78E-10	6.16E-10	4.37E-10	3.16E-10
19000	4.08E-10	1.45E-09	1.99E-09	1.74E-09	1.27E-09	8.89E-10	6.22E-10	4.41E-10	3.19E-10
20000	4.55E-10	1.53E-09	2.06E-09	1.78E-09	1.29E-09	9.00E-10	6.28E-10	4.45E-10	3.21E-10

Table 3. Calculated Values of Coefficient K_{ci} [cm³/s] as a function of n and T ($2 \le n \le 10$ and $3000 \text{ K} \le T \le 20000 \text{ K}$). The values for $2 \le n \le 8$ and $4000 \text{ K} \le T \le 10000 \text{ K}$ are from Mihajlov et al. (2011) and here are given for integrity.

Extended calc.

Т

n

T/n	10	11	12	13	14	15	16	17	18	19	20
3000	1.92E-10	1.51E-10	1.19E-10	9.44E-11	7.56E-11	6.11E-11	4.98E-11	4.09E-11	3.39E-11	2.82E-11	2.37E-11
4000	2.17E-10	1.68E-10	1.31E-10	1.03E-10	8.19E-11	6.57E-11	5.32E-11	4.35E-11	3.58E-11	2.97E-11	2.49E-11
5000	2.36E-10	1.81E-10	1.40E-10	1.09E-10	8.62E-11	6.88E-11	5.55E-11	4.52E-11	3.72E-11	3.08E-11	2.57E-11
6000	2.51E-10	1.90E-10	1.46E-10	1.14E-10	8.94E-11	7.12E-11	5.73E-11	4.65E-11	3.82E-11	3.16E-11	2.63E-11
7000	2.62E-10	1.98E-10	1.51E-10	1.17E-10	9.20E-11	7.30E-11	5.86E-11	4.76E-11	3.89E-11	3.22E-11	2.68E-11
8000	2.71E-10	2.04E-10	1.55E-10	1.20E-10	9.40E-11	7.45E-11	5.97E-11	4.84E-11	3.96E-11	3.26E-11	2.72E-11
9000	2.79E-10	2.09E-10	1.59E-10	1.22E-10	9.57E-11	7.57E-11	6.06E-11	4.90E-11	4.01E-11	3.30E-11	2.75E-11
10000	2.85E-10	2.13E-10	1.62E-10	1.24E-10	9.71E-11	7.67E-11	6.14E-11	4.96E-11	4.05E-11	3.34E-11	2.77E-11
11000	2.91E-10	2.17E-10	1.64E-10	1.26E-10	9.83E-11	7.76E-11	6.20E-11	5.01E-11	4.09E-11	3.36E-11	2.79E-11
12000	2.96E-10	2.20E-10	1.66E-10	1.28E-10	9.93E-11	7.83E-11	6.25E-11	5.05E-11	4.12E-11	3.39E-11	2.81E-11
13000	3.00E-10	2.23E-10	1.68E-10	1.29E-10	1.00E-10	7.90E-11	6.30E-11	5.09E-11	4.14E-11	3.41E-11	2.83E-11
14000	3.04E-10	2.25E-10	1.70E-10	1.30E-10	1.01E-10	7.96E-11	6.35E-11	5.12E-11	4.17E-11	3.43E-11	2.84E-11
15000	3.08E-10	2.28E-10	1.71E-10	1.31E-10	1.02E-10	8.01E-11	6.38E-11	5.15E-11	4.19E-11	3.44E-11	2.86E-11
16000	3.11E-10	2.30E-10	1.73E-10	1.32E-10	1.02E-10	8.06E-11	6.42E-11	5.17E-11	4.21E-11	3.46E-11	2.87E-11
17000	3.14E-10	2.31E-10	1.74E-10	1.33E-10	1.03E-10	8.10E-11	6.45E-11	5.19E-11	4.23E-11	3.47E-11	2.88E-11
18000	3.16E-10	2.33E-10	1.75E-10	1.34E-10	1.04E-10	8.14E-11	6.48E-11	5.21E-11	4.24E-11	3.48E-11	2.89E-11
19000	3.19E-10	2.35E-10	1.76E-10	1.34E-10	1.04E-10	8.17E-11	6.50E-11	5.23E-11	4.26E-11	3.50E-11	2.90E-11
20000	3.21E-10	2.36E-10	1.77E-10	1.35E-10	1.04E-10	8.20E-11	6.52E-11	5.25E-11	4.27E-11	3.51E-11	2.90E-11

In order to enable the better and more adequate use of data, we give for the rate coefficients a simple and accurate <u>fitting formula</u> based on a least-squares method, which is logarithmic and represented by a second-degree polynomial

$$\log(K_{\rm ci}(T)) = k_1 + k_2 \log(T) + k_3 (\log(T))^2.$$

Table 2. The fits of the equation (13) to the rate coefficient. A portion is shown here for guidance regarding its form.

n	k_1	k_2	k_3
4	19.91758	- 15.14785	1.98009
5	-20.60455	5.21174	- 0.57122
6	-17.85054	3.94548	- 0.43315
7	-16.16158	3.1383	- 0.34519
8	- 15.03989	2.57737	-0.28382
9	-14.20649	2.14118	-0.23553
10	- 13.64156	1.82851	-0.20101
11	- 13.26126	1.60326	- 0.17639
12	- 12.96479	1.41629	-0.1557
13	- 12.75919	1.27432	-0.14016
14	- 12.62833	1.1705	- 0.12912
15	-12.4765	1.05481	- 0.11608
16	-12.37305	0.96497	- 0.10616
17	-12.31258	0.89712	-0.09875
18	- 12.24943	0.82884	- 0.09113
19	-12.19548	0.76602	-0.08407
20	-12.16014	0.71392	-0.07827

The fits are valid over the temperature range of $2000K \le T \le 20\ 000\ K$. Also, it is possible that the fit is applicable outside this area but with caution. In the Table 2, the selected fits are presented (for $4 \le n \le 20$).

 \sim

 $A^{*}(n) + A \Longrightarrow A_{2}^{+} + e \qquad (4)$ $A^{*}(n) + A \Longrightarrow A + A^{+} + e \qquad (5)$ $A_{2}^{+} + e \implies A^{*}(n) + A \qquad (6)$ $A + A^{+} + e \Longrightarrow A^{*}(n) + A \qquad (7)$

Concurrent processes

$$A^*(n) + e \Rightarrow A^+ + 2e,$$

$$A^+ + 2e \Rightarrow A^*(n) + e,$$

$$A^+ + e \Rightarrow A^*(n) + \varepsilon_{\lambda},$$

 $I_{ci}(n, T) = K_{ci}(n, T) \cdot N_n N_1,$ $I_{cr}(n, T) = K_{cr}(n, T) \cdot N_1 N_i N_e,$

chemi-ionization/recombination fluxes



The influence of analyzed processes increases linearly with N, and, for example for $N = 10^{13}$ cm⁻³ some optical characteristics may be different than for $N = 10^{12}$ cm⁻³.

vrdnik, Serbia Ratio of fluxes define the importance of CI

Namely, the influence of analyzed processes increases linearly with N, and, for example for $N = 10^{13}$ cm⁻³ some optical characteristics may be different than for $N = 10^{12}$ cm⁻³, for example due to changes in energy level populations, electron density, influence on the formation of hydrogen molecule, opacity, line profiles etc.

We can see as well that even around the value of densities <u>10¹² cm⁻³</u> the inclusion of the considered here chemi-ionization/recombination processes <u>could improve</u> the modelling and analysis of such regions not only in photospheres of Sun and solar like stars but also in clouds in AGN BLR.

Additionally, the Figure 3 demonstrates as well the high <u>sensitivity</u> of the influence of these processes to the <u>relatively small changes of densities</u> which can be of interest for the determination of limiting densities in clouds in AGN BLR.

It could be very useful to perform an analysis for example with the code CLOUDY in order to see which changes in optical characteristics may be used in order to establish the presence of such dense layers.

The results show that the considered chemi ionization/ recombination processes could be <u>used for determination of limiting high</u> densities in clouds in AGN BLR region and for the improvement of modelling of dense moderately ionized layers in them.

If exist changes in optical charact. (with inclusion ch.ion.) =>

than there exists high densities in clouds in AGN BLR region

To conclude

The obtained results demonstrate the fact that the considered <u>chemi-ionization/recom</u>bination processes, which <u>influence</u> on the <u>ionization</u> level and atom <u>excited-state populations</u>, must have a very <u>significant influence</u> on the <u>optical properties</u> of the weakly ionized regions where the neutral <u>hydrogen densities are larger than 10¹² cm⁻³</u> since in such conditions they dominate over the relevant concurrent electron–atom collision processes.

The possibility that the chemi-ionization processes in atom–Rydberg atom collisions, as well as the corresponding chemi-recombination processes, may be <u>useful</u> for the <u>diagnostics</u>, modelling, and confirmation of existence or non-existence of very dense weakly ionized domains in clouds in broad-line region of active galactic nuclei, has been considered.

Data are also useful for stellar plasma investigation:

This can be used as a diagnostic method to find out if the domains with such densities exist or not. Additionally, our previous results obtained for principal quantum number $2 \le n \le 8$ and $4000K \le T \le 10\ 000K$ are <u>extended</u> for principal quantum number $9 \le n \le 20$ and $10\ 000K < T \le 20\ 000K$ and also for low-temperature region ($T < 4000\ K$) for $2 \le n \le 20$. The results obtained during this investigation are in the process of inserting in our MoID molecular database which is web services at the Serbian virtual observatory (SerVO) and nodes within the Virtual Atomic and Molecular Data Center (VAMDC).



- VAMDC is consortium for sharing Atomic and Molecular data
- Currently 33 databases running in the consortium

ALADDIN2, BASECOL, Belgrade electron/atom(molecule) database (BEAMDB), CDMS, Carbon Dioxide Spectroscopic Databank 1000K, Carbon Dioxide Spectroscopic Databank 296K, Chianti, DESIRE, ECaSDa, S&MPO, GhoSST, <u>HITRANonline</u>, IDEADB, JPL database, KIDA, LXcat, MeCaSDa, OACT – LASP Database, PAH, Photodissociation - MolD database, <u>RADAM</u> - Ion Interactions, SHeCaSDa, SpEctroScopy of Atoms and Molecules Sesam, Spectr-W3, <u>Stark-B</u>, TFMeCaSDa, TIPbase, TOPbase, UMIST Database for Astrochemistry, <u>VALD</u> (atoms), VALD subset in Moscow, VAMDC species-DB, Water, ...

• Large diversity among data

spectral line lists of atoms and molecules

transition probabilities

cross-sections (ro-vibrational, photodissociation, electron interaction...)

kinetic data

reaction rate coefficients

« Stark » broadening

VAMDC portal

<u>http://portal.vamdc.org/vamdc_portal/home.seam</u>





Map of databases (on all continents)

THANK YOU FOR ATTENTION



12th SCSLSA 3-7. June 2019 Vrdnik, Serbia

12th Serbian Conference on Spectral Line Shapes in Astrophysics

<u>The (n –n')-mixing processes in the Broad Line</u> <u>Region of AGNs</u>

Also in collisions H(n) + H(1s) i.e. Atom–Rydberg atom collisions we need to investigate excitation/de-excitation processes.

Our Future work => our current work. Next time talk in details.

excitation processes

$$\mathbf{H}^{*}(n) + \mathbf{H} \to \begin{cases} \mathbf{H}^{*}(n' = n + p) + \mathbf{H}, \\ \mathbf{H} + \mathbf{H}^{*}(n' = n + p), \end{cases}, \qquad n \ge 4, \quad p \ge 1, \ (1)$$

and the inverse process of de-excitation

 $\mathbf{H}^*(n) + \mathbf{H} \rightarrow \begin{cases} \mathbf{H}^*(n' = n - p) + \mathbf{H}, \\ \mathbf{H} + \mathbf{H}^*(n' = n - p), \end{cases}$

$$n-p \ge 4, \tag{2}$$



Table 2. Excitation rate coefficients $K_{\pi\pi+p}(T)$ (10^{-9} cm³s⁻¹). The values for $4 \le n \le 10$ and 3000 K $\le T \le 7000$ K are from Mihajlov et al. (2004) and here are given for integrity.

						Т	[K]				
n	Р	2000	3000	5000	7000	8000	9000	10000	15000	20000	25000
	1	2.82409	4.42323	6.24561	7.18872	7.50148	7.74978	7.95119	8.56558	8.87458	9.05872
	2	0.32878	0.72276	1.32049	1.68755	1.81762	1.92380	2.01179	2.29039	2.43614	2.52470
4	3	0.09004	0.24014	0.50917	0.69174	0.75905	0.81493	0.86185	1.01388	1.09539	1.14553
	4	0.03617	0.10906	0.25438	0.35948	0.39924	0.43261	0.46087	0.55383	0.60448	0.63590
	5	0.01806	0.05920	0.14738	0.21397	0.23960	0.26128	0.27975	0.34114	0.37496	0.39606
	1	2.68161	3.37083	4.00610	4.29430	4.38513	4.45573	4.51208	4.67927	4.76096	4.80898
	2	0.50323	0.76422	1.04763	1.18951	1.23592	1.27255	1.30213	1.39164	1.43627	1.46275
5	3	0.17648	0.30197	0.45344	0.53427	0.56137	0.58299	0.60057	0.65453	0.68182	0.69812
	4	0.08211	0.15238	0.24354	0.29437	0.31171	0.32563	0.33702	0.37231	0.39034	0.40116
	5	0.04510	0.08870	0.14824	0.18250	0.19433	0.20388	0.21172	0.23618	0.24877	0.25636
	1	1.97872	2.24840	2.47312	2.56897	2.59849	2.62123	2.63924	2.69208	2.71760	2.73253
	2	0.47651	0.60719	0.72780	0.78249	0.79972	0.81311	0.82380	0.85548	0.87096	0.88005
6	3	0.19300	0.26584	0.33798	0.37211	0.38304	0.39159	0.39844	0.41894	0.42904	0.43500
	4	0.09850	0.14346	0.19031	0.21317	0.22057	0.22639	0.23107	0.24517	0.25215	0.25629
	5	0.05765	0.08753	0.11985	0.13598	0.14125	0.14541	0.14876	0.15891	0.16396	0.16696
	1	1.36415	1.47213	1.55736	1.59254	1.60324	1.61145	1.61792	1.63682	1.64590	1.65120
	2	0.38004	0.44152	0.49348	0.51583	0.52273	0.52805	0.53227	0.54466	0.55066	0.55417
7	3	0.16853	0.20645	0.24012	0.25503	0.25969	0.26330	0.26617	0.27464	0.27875	0.28117
	4	0.09148	0.11660	0.13973	0.15020	0.15350	0.15606	0.15811	0.16417	0.16712	0.16886
	5	0.05597	0.07354	0.09018	0.09785	0.10028	0.10217	0.10368	0.10818	0.11038	0.11168

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The (n - n')-mixing processes in the Broad Line Region of AGNs: data needed for spectroscopy diagnostics

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e-mail: vlada@ipb.ac.rs Received 06/11/2017; accepted 09/01/2018 The aim of our <u>further investigation</u> is to study the physics of AGN, i.e. to investigate the <u>atomic processes</u> (collisional atom - Rydberg atom i.e. chemi-ionization/recombination and n-n' mixing processes) and <u>revise</u> their <u>role</u>.

This means to find out at what plasma conditions certain atomic processes become important and could explain the existence of AGN regions with such characteristics, and could be used for future diagnostics, numerical simulations and modelling.

We will concentrate on:

- Calculations of the <u>Balmer line ratios</u> including chemiionization/recombination and n-n' mixing processes.

- Selection of the spectra of AGNs with broad emission lines in their spectra (type 1 AGN) where we expect that this process is dominant. The <u>sample</u> will contain more than 100 AGNs with high signal-to-noise ratio, from different databases (as e.g. SDSS database)

- Using different spectral tools we will <u>fit</u> the contininuum and broad emission lines (including Fe II optical broad lines) in order to get pure broad Balmer lines

- <u>Comparison</u> of the calculated and observed Balmer line ratios in order to extract the influence of chemi-ionization/recombination and n-n' mixing processes in the BLR.



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research interest:

Solar and stellar astrophysics; High energy astrophysics; Atomic and ionic collisions with formation of quasimolecules; Atomic processes in white dwarfs and solar type stars; Astroinformatics; Databases; Space Weather studies of Upper Atmosphere; Ionospheric plasma Irregularities using VLF.

Also, people from the IF team: Lj.M. Ignjatović, A. Nina, N.Sakan ...

AOB team i.e. collaborators in this thematic: Milan S. Dimitrijević, and othersDarko Jevremović, Luka Popović, Zoran Simić, Veljko Vujčić

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collisional ionization